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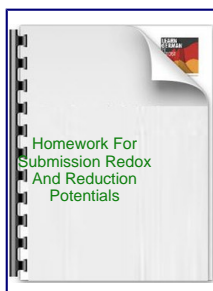


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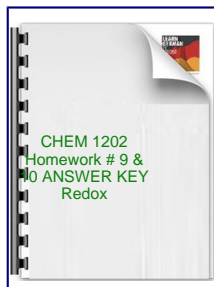


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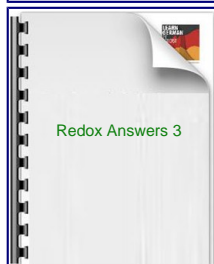
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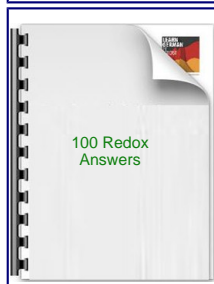
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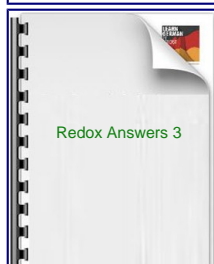
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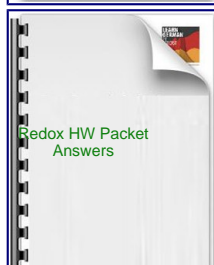
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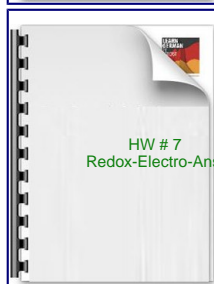
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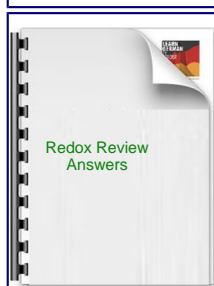
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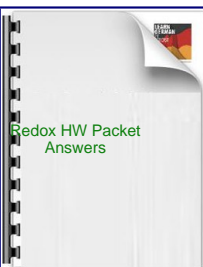
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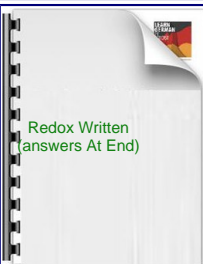
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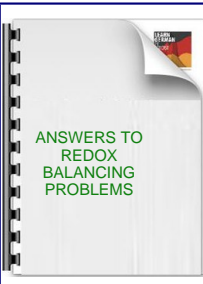
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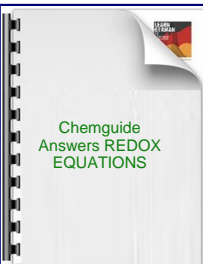
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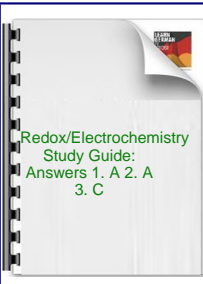
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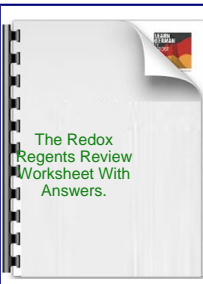
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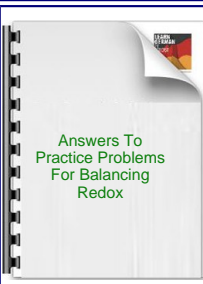
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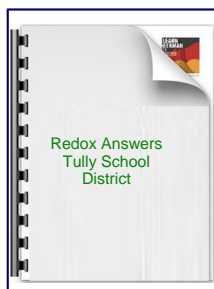
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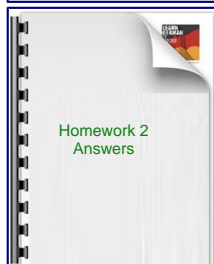
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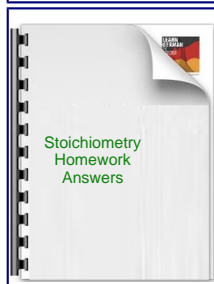
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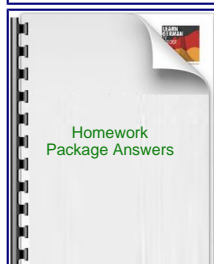
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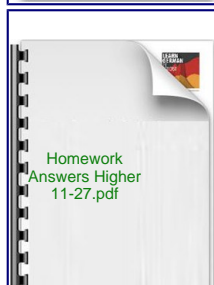
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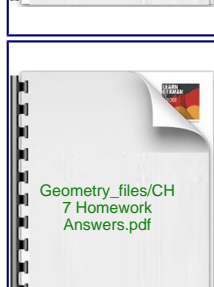
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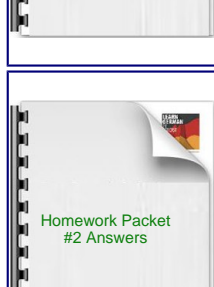
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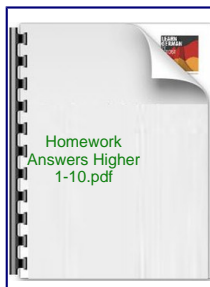
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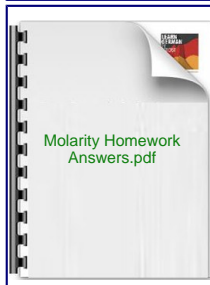
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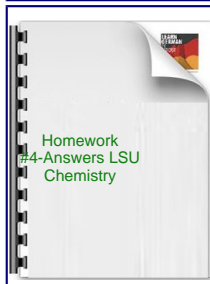
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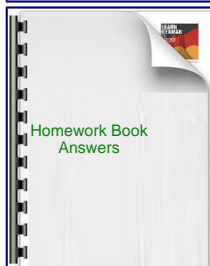
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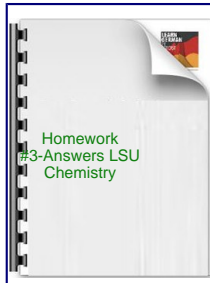
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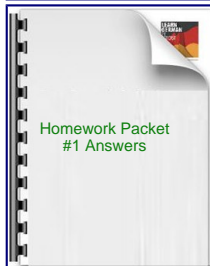
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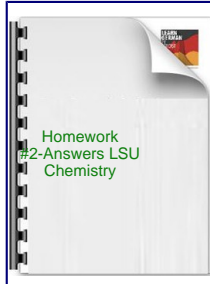
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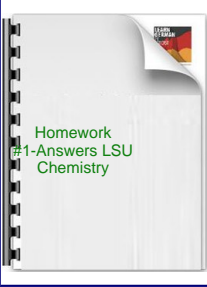

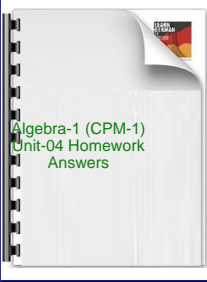

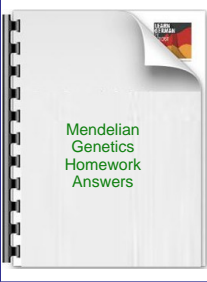

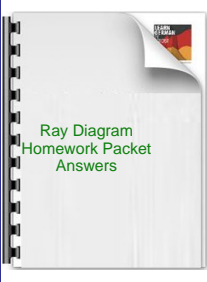

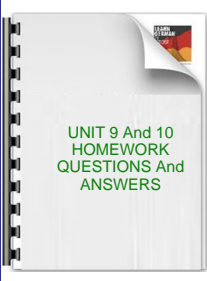

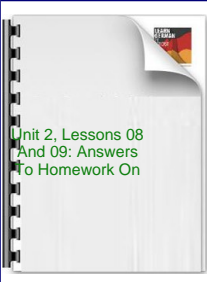

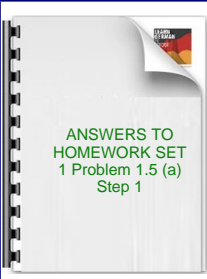

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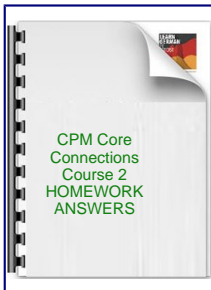


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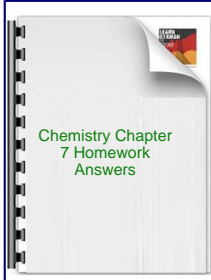


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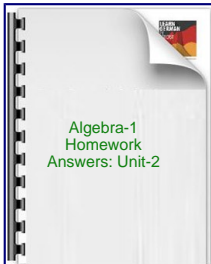
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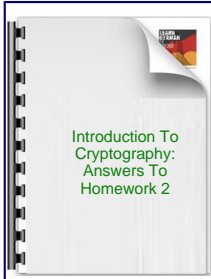
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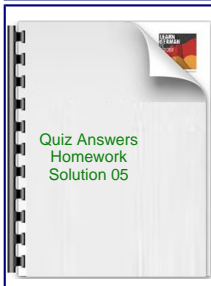
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Chem 142B, HW 5 answer key, Fall 2005. Solution: This is going to be a redox reaction, since we're going to oxidize the chromium involved. First, I. This PDF book provide homework oxidation and reduction answer key document. To download free quiz answers homework solution 05 you need to register.

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